

Intramolecular Cyclizations of Enynes Using $\text{RuClH}(\text{CO})(\text{PPh}_3)_3$

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Transition metal-catalyzed coupling reaction between alkyne and alkene is a useful synthetic protocol for synthesis of 1,3-diene.¹ Not only are 1,3-dienes themselves important but they have also wide applicability in the Diels–Alder reaction. Recently, Mitsudo² and Trost³ reported ruthenium-catalyzed intermolecular addition of alkynes to alkenes. In both reactions, formation of ruthenacyclopentene was proposed as an intermediate. We now report two types of intramolecular cyclizations of enynes using $\text{RuClH}(\text{CO})(\text{PPh}_3)_3$ as a catalyst via the following three steps: hydro-metalation, carbometalation, and then β -hydride elimination,⁴ which provide cyclized compounds **I**⁵ and **II**, respectively (Scheme 1). Among the many reports on hydorruthenation of $\text{RuClH}(\text{CO})(\text{PPh}_3)_3$ toward multiple bonds,⁶ this is the first example of a stereoselective carbon–carbon bond-forming reaction using $\text{RuClH}(\text{CO})(\text{PPh}_3)_3$.

First, we examined ruthenium-catalyzed cyclizations using **1a** as a substrate ($n = 3$, Table 1, run 1). A solution of **1a** (0.32 mmol) and $\text{RuClH}(\text{CO})(\text{PPh}_3)_3$ (5 mol %) in toluene (1.5 mL) was refluxed for 9 h to afford the cyclized product **2a** in 62% yield. A strong NOE between two olefinic protons of **2a** clarified the structure of the product. The cyclization of **1b** also succeeded in providing **2b** in 57% yield (run 2). It is notable that both substrates, having an aromatic group or an alkyl group on the alkyne, can be used in this reaction.⁷

To examine the substituent effects on the aromatic ring, the cyclization of **1c**, having a methoxy group on the aromatic ring, was carried out to provide **2c** in 82% yield (run 3). Similarly, the reaction of **1d**, having a methyl group, proceeded to give **2d** in 67% yield (run 4). However, in the case of **1e**, having a trifluoromethyl group, the cyclized product **2e** was obtained in 53% yield along with recovered starting material **1e** (34% yield) (run 5). These results clearly indicated that the electron-withdrawing group on the alkyne reduced the yield of the product.

Surprisingly, when cyclization of **3a**, having a two-carbon tether between the alkyne and olefin, was carried out, the reaction was accomplished within 1 h to afford the cyclized

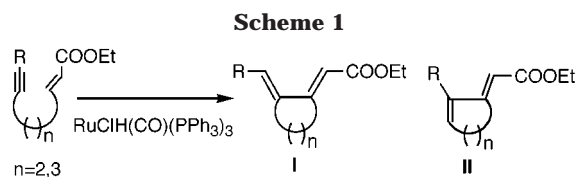
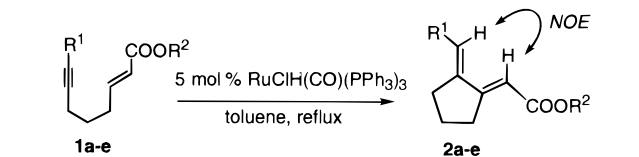
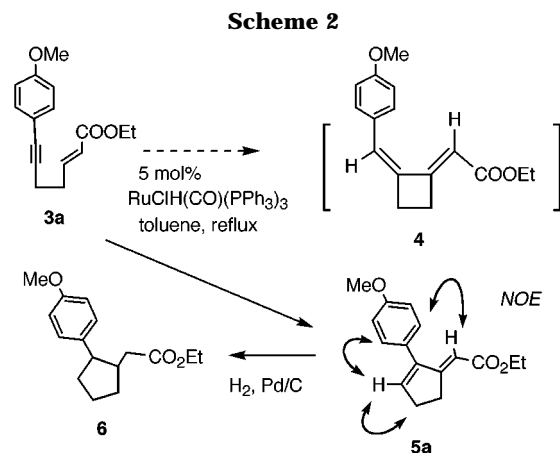


Table 1. Cyclization of **1** Using $\text{RuClH}(\text{CO})(\text{PPh}_3)_3$



run	substrate	R ₁	R ₂	h	2a–e (%)
1	1a	Me	(CH ₂) ₃ Ph	9	62
2	1b	Ph	Et	18	57
3	1c	4-MeO-Ph	Et	18	82
4	1d	4-Me-Ph	Et	18	67
5	1e	4-CF ₃ -Ph	Et	18	53 ^a

^a The recovery of **1e** was 34%.



product in 81% yield (Scheme 2). However, ¹H NMR and ¹³C NMR spectra could not clarify the ring size of the product (**4** or **5a**). To confirm the structure of **5a**, hydrogenation was carried out to give **6** as a mixture of two isomers. The COSY data of one isomer confirmed that a five-membered ring was formed in this reaction.⁸

The effects of substituents on the aromatic ring were studied again (Table 2). Cyclization of **3b** afforded **5b** in 75% yield (run 2), and cyclization of **3c** provided **5c** in 67% yield (run 3). However, the reaction of **3d** was not completed to provide **5d** in only 48% yield along with **3d** (run 4). Accordingly, a tendency similar to that shown in the previous cyclization was also observed in this reaction.⁹

The reaction mechanism can be envisioned as shown in Scheme 3. The reaction starts with a hydorruthenation of the alkyne to give the vinylruthenium complex **III** or **IV**, which is in a state of equilibrium with **1** or **3**. In the reaction of **1** and $\text{RuClH}(\text{CO})(\text{PPh}_3)_3$, intramolecular olefin insertion

(8) HMCB data of **5a** indicated that the aromatic group was connected to the tertiarily vinylic carbon.

(9) Cyclizations were examined using **3e** (R¹ = Me, R² = 4-NO₂-C₆H₄-CH₂), **3f** (R¹ = Et, R² = 4-NO₂-C₆H₄CH₂), and **3g** (R¹ = *i*-Pr, R² = Et) as the substrates. In the reaction of **3e**, the starting material was recovered. In the case of **3g**, double-bond isomerization took place to provide deconjugated compounds. However, in the reaction of **3f**, a cyclized product was obtained in 22% yield along with 15% of **3f**.

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(2) Mitsudo, T.; Zhang, S.; Naagao, M.; Wakatuki, Y. *J. Chem. Soc., Chem. Commun.* **1991**, 598.

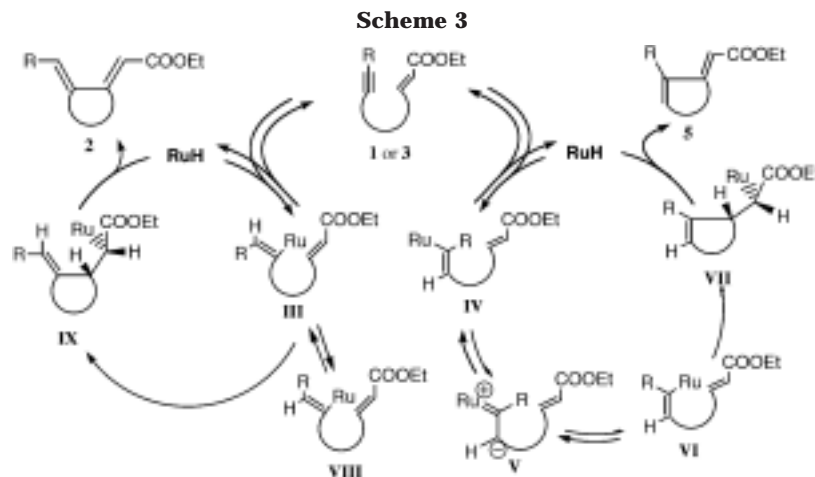
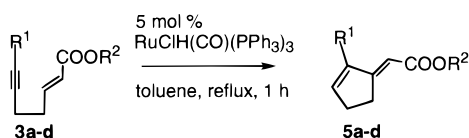
(3) (a) Trost, B. M.; Katsuharu Imi; Indolese, A. *J. Am. Chem. Soc.* **1993**, *115*, 8831. (b) Trost, B. M.; Indolese, A. F.; Müller, T. J. J.; Treptow, B. *J. Am. Chem. Soc.* **1995**, *117*, 615. (c) Trost, B. M.; Müller, T. J. J.; Martinez, J. *J. Am. Chem. Soc.* **1995**, *117*, 1878.

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(7) Cyclizations were examined using **1f** (R¹ = H, R² = (CH₂)₃Ph), **1g** (R¹ = TMS, R² = (CH₂)₃Ph), and **1h** (R¹ = COOMe, R² = COOEt) as substrates. In the case of **1f** and **1g**, double-bond isomerization took place to provide deconjugated compounds. In the reaction of **1h**, the starting material was recovered. Unfortunately, a six-membered ring compound was not formed in the reaction of (*E*)-9-(4-methoxyphenyl)-2-nonen-7-yne.

**Table 2. Cyclization of 3 Using RuClH(CO)(PPh₃)₃**

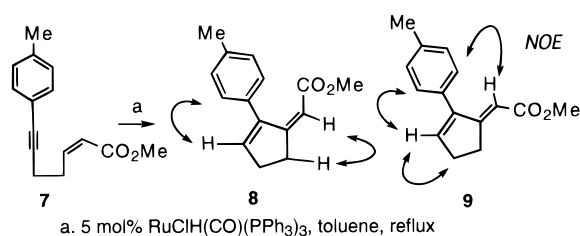
run	substrate	R ¹	R ²	5a-d (%)
1	3a	4-MeO-Ph	Et	81
2	3b	4-Me-Ph	Et	75
3	3c	Ph	Et	67
4	3d	4-CF ₃ -Ph	Et	48 ^a

^a The recovery of **3d** was 48%.

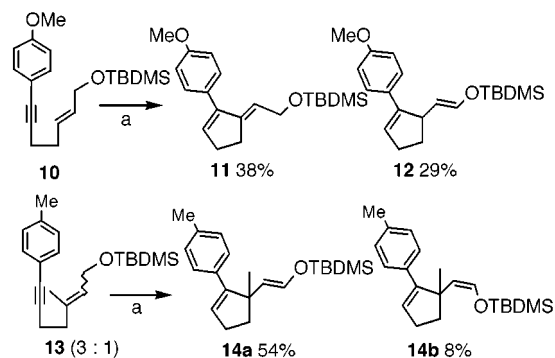
into the Ru–C bond of **III** proceeds and is followed by a syn β -H elimination to produce the cyclized product **2** stereoselectively. On the other hand, in the reaction of **3**, after vinylruthenium complex **IV** (a cis addition product) isomerized to vinylruthenium complex **VI** (via a dipolar intermediate **V**),¹⁰ intramolecular olefin insertion took place, and successive syn β -H elimination produced **5**, exclusively. In both reactions of **1** and **3**, five-membered ring formation preceded all other ring formations.¹¹ A remarkable characteristic in our reaction is the formation of a cyclopentene moiety conjugated with an *exo*-olefin.

Next, the cyclization of substrate **7** (*Z* isomer of **3b**) was examined (Table 3). After 20 min, cyclized products **8** and **9** were obtained in 33% and 10% yields, respectively, along with 22% of the recovered starting material. NOE experiments clarified the structures of **8** and **9**. The major product **8** had the expected stereochemistry with regard to the α,β -unsaturated ester. The prolonged reaction time (1 h) did not increase the yield of **8** (30%), although it increased the yield of **9** (25%). When **8** was treated under the same reaction conditions, isomerization of **8** into **9** occurred.

Finally, we examined the cyclization of **10**, which does not have an electron-withdrawing group on the alkene functionality. The reaction of **10** with ruthenium complex gave **11** (38% yield) and **12** (29% yield) (Scheme 4). It is notable that intramolecular olefin insertion took place on the alkene without the presence of an electron-withdrawing group.

Table 3. Cyclization of 7 (*Z* Isomer of 3b)

run	time	yields (%)		
		8	9	7
1	20 min	33	10	22
2	1 h	30	25	

Scheme 4^a

^a Key: (a) 5 mol % RuClH(CO)(PPh₃)₃, toluene, reflux, 1 h.

When **13** was treated in a similar manner, the products were only silyl enol ethers **14a** (54% yield) and **14b** (8% yield), and *E* isomer **14a** was the major product.

In summary, we have succeeded in the first example of an intramolecular cyclization of an enyne substrate using RuClH(CO)(PPh₃)₃ as a catalyst. Formation of cyclopentene derivatives is a prominent characteristic in our reaction (Scheme 1). Further studies on cyclizations using ruthenium catalysts are in progress.

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Supporting Information Available: Experimental details and characterization for **1b–e**, **2a–e**, **3a–d**, **5a–d**, **6–13**, and **14a,b** (12 pages).

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(11) Since the same tendency in the effects of substituents on the aromatic ring was shown in each reaction (Tables 1 and 2), the rate-determining step would be irreversible intramolecular olefin insertion. An electron-withdrawing group on alkyne stabilized the produced vinylruthenium complex through back-donation, which retarded olefin insertion.